

Data collection

Rigaku AFC-5 diffractometer	$R_{\text{int}} = 0.024$
ω scans	$\theta_{\text{max}} = 27.51^\circ$
Absorption correction: none	$h = -12 \rightarrow 12$
3704 measured reflections	$k = -12 \rightarrow 2$
1821 independent reflections	$l = -11 \rightarrow 11$
1571 reflections with $I > 2\sigma(I)$	3 standard reflections every 200 reflections
	intensity decay: none

Refinement

Refinement on F^2	$(\Delta/\sigma)_{\text{max}} < 0.001$
$R[F^2 > 2\sigma(F^2)] = 0.036$	$\Delta\rho_{\text{max}} = 0.302 \text{ e } \text{\AA}^{-3}$
$wR(F^2) = 0.100$	$\Delta\rho_{\text{min}} = -0.206 \text{ e } \text{\AA}^{-3}$
$S = 1.113$	Extinction correction: none
1821 reflections	Scattering factors from International Tables for Crystallography (Vol. C)
128 parameters	
All H atoms refined	
$w = 1/[\sigma^2(F_o^2) + (0.0509P)^2 + 0.1496P]$	
where $P = (F_o^2 + 2F_c^2)/3$	

Table 1. Hydrogen-bonding geometry (\AA , $^\circ$)

D—H...A	D—H	H...A	D...A	D—H...A
N1—H1...S ⁱ	0.83 (3)	2.69 (2)	3.481 (2)	162 (1)
N2—H22...S ⁱⁱ	0.85 (3)	2.50 (2)	3.337 (2)	169 (1)
N2—H21...C5 ⁱⁱⁱ	0.83 (3)	2.74 (3)	3.318 (3)	128 (2)

Symmetry codes: (i) $x, \frac{1}{2}-y, z-\frac{1}{2}$; (ii) $x, \frac{1}{2}-y, \frac{1}{2}+z$; (iii) $x, -\frac{1}{2}-y, \frac{1}{2}+z$.

Program(s) used to solve structure: *SHELXS96* (Sheldrick, 1996a). Program(s) used to refine structure: *SHELXL96* (Sheldrick, 1996b). Molecular graphics: *PLUTON* (Spek, 1990). Software used to prepare material for publication: *SHELXL96*.

The author thanks the MINERVA Foundation, Munich, Germany, for awarding a fellowship at the Weizmann Institute, Rehovot, Israel, in the laboratory of Professor Joel L. Sussman. Dr Linda Shimon is thanked for helping with the X-ray data collection.

Supplementary data for this paper are available from the IUCr electronic archives (Reference: JZ1278). Services for accessing these data are described at the back of the journal.

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Acta Cryst. (1998). **C54**, 1123–1125

2',4'-Dihydroxy-2-[3-(4-methylphenyl)-isoxazol-5-yl]acetophenone†

NARESH KUMAR,^a SUCHARITA ROY,^a VIRINDER S. PARMAR^a AND WILLIAM ERRINGTON^b

^aDepartment of Chemistry, University of Delhi, Delhi 110 007, India, and ^bDepartment of Chemistry, University of Warwick, Coventry CV4 7AL, England. E-mail: w.errington@warwick.ac.uk

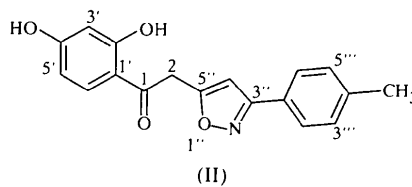
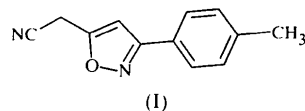
(Received 12 January 1998; accepted 19 February 1998)

Abstract

The synthesis and structure of the title compound, $\text{C}_{18}\text{H}_{15}\text{NO}_4$, are reported. The dihydroxyphenyl and methylphenyl groups are inclined at angles of 84.98 (4) and 9.18 (10)°, respectively, to the plane of the isoxazole ring. Both intra- and intermolecular hydrogen-bonding interactions are identified.

Comment

Due to the known various biological activities of isoxazoles (an important example being antispasmodic; Naruto *et al.*, 1982, 1983), we have synthesized a series of isoxazoles for further testing. Recently, we reported (Kumar *et al.*, 1996) the synthesis and X-ray structure of [3-(4-methylphenyl)-5-isoxazolyl]acetonitrile, (I); in the process of its derivatization (for structure–activity studies), we condensed it with resorcinol under Hoesch reaction conditions to produce the corresponding β -res-acetophenone derivative (II). Such compounds have not been reported previously. This X-ray investigation was undertaken in order to unambiguously assign its constitution.



† IUPAC name: 2,4-dihydroxyphenyl 3-(4-methylphenyl)isoxazol-5-ylmethyl ketone.

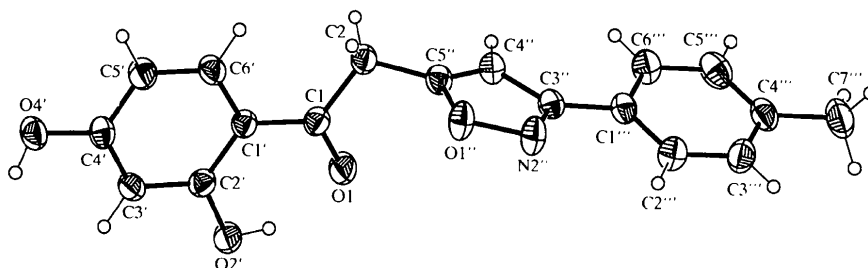


Fig. 1. View of the title molecule showing the atomic numbering. Displacement ellipsoids are drawn at the 50% probability level.

The structure of the title compound is illustrated in Fig. 1. The bond lengths and angles are unexceptional (Allen *et al.*, 1987). There is intramolecular hydrogen bonding between atoms O1 and O2' [O1...O2' 2.5493 (14) Å], and this interaction is presumably responsible for the approximately planar alignment of the O1—C1—C2 unit with respect to the plane of the dihydroxyphenyl group [O1—C1—C1'—C6' −178.22 (14)°]. The isoxazole ring is essentially flat, with an r.m.s. deviation of 0.0017 Å. The dihydroxyphenyl and methylphenyl groups are inclined at angles of 84.98 (4) and 9.18 (10)°, respectively, to the mean plane through the isoxazole group. Structural stability is enhanced by intermolecular hydrogen bonding between the O4' and N2''(−*x*, *y* + ½, −*z* + ½) atoms [N...O 2.8124 (16) Å].

Experimental

[3-(4-Methylphenyl)-5-isoxazolyl]acetonitrile [0.99 g, 5 mmol; (I)] and resorcinol (0.55 g, 5 mmol) were dissolved in dry ether (75 ml), the solution cooled in an ice-salt mixture and powdered anhydrous ZnCl₂ added. A rapid stream of dry HCl gas was passed through the mixture for 2 h with occasional shaking. The mixture was kept in an ice chest for 24 h and dry HCl gas was then passed through it for a further 2 h. After the mixture was maintained at 278 K for a further 2 d, the clear ethereal solution was decanted and the dark-brown sticky solid was washed twice with dry ether. The brown solid was dissolved in hot water (125 ml) and the solution refluxed for 2 h, whereupon a light-brown solid precipitated out. This was filtered, washed with water and then with petroleum ether, and recrystallized from acetone-petroleum ether as pale-pink needles (0.85 g, 55% yield; m.p. 458 K). IR (Nujol): 3180 (OH), 1660 (C=O), 1362, 1236, 1175, 1150, 820 and 805 cm^{−1}; ¹H NMR (90 MHz, acetone-*d*₆): 2.23 (3H, *s*, C-4'''-CH₃), 3.82 (1H, *bs*, C-4'-OH), 4.50 (2H, *s*, C-2H), 6.40 (2H, *m*, C-3'H and C-5'H), 6.68 (1H, *s*, C-4''H), 7.20 (2H, *d*, *J* = 8 Hz, C-3''H and C-5''H), 7.70 (3H, *m*, C-2''H, C-6''H and C-6'H) and 12.10 (1H, *bs*, C-2'-OH); EIMS, *m/z* (% relative intensity): 309 [*M*⁺] (40), 218 (10), 217 (50), 200 (12), 173 (40), 172 (10), 158 (80), 137 (85), 115 (20), 109 (12), 91 (100), 81 (55) and 65 (60).

Crystal data

C₁₈H₁₅NO₄
M_r = 309.31
 Monoclinic
*P*2₁/*c*
a = 7.9188 (14) Å
b = 15.010 (3) Å
c = 13.422 (2) Å
 β = 101.064 (3)°
V = 1565.7 (5) Å³
Z = 4
D_x = 1.312 Mg m^{−3}
D_m not measured

Mo K α radiation

λ = 0.71073 Å
 Cell parameters from 4695 reflections
 θ = 2.06–28.53°
 μ = 0.093 mm^{−1}
T = 180 (2) K
 Block
 0.50 × 0.24 × 0.20 mm
 Colourless

Data collection

Siemens SMART CCD area-detector diffractometer
 ω scans
 Absorption correction: none
 9463 measured reflections
 3667 independent reflections
 2462 reflections with *I* > 2 σ (*I*)

*R*_{int} = 0.028
 θ_{\max} = 28.53°
h = −9 → 10
k = −19 → 14
l = −17 → 17
 Intensity decay: none

Refinement

Refinement on *F*²
R [*F*² > 2 σ (*F*²)] = 0.043
wR(*F*²) = 0.111
S = 0.990
 3667 reflections
 213 parameters
 H atoms riding
 $w = 1/[\sigma^2(F_o^2) + (0.0581P)^2]$
 where $P = (F_o^2 + 2F_c^2)/3$

(Δ/σ)_{max} = 0.001
 $\Delta\rho_{\max}$ = 0.211 e Å^{−3}
 $\Delta\rho_{\min}$ = −0.202 e Å^{−3}
 Extinction correction: none
 Scattering factors from
International Tables for Crystallography (Vol. C)

Table 1. Selected geometric parameters (Å, °)

O1''—C5''	1.3530 (16)	N2''—C3''	1.3160 (17)
O1''—N2''	1.4064 (15)		
N2''—C3''—C4''	110.52 (12)	C5''—C4''—C3''	105.35 (12)
N2''—C3''—C1'''	119.85 (12)	C4''—C5''—O1''	109.71 (12)
O1—C1—C1'—C6'	−178.22 (14)	C4''—C3''—C1'''—C6'''	8.4 (2)
C2—C1—C1'—C2'	−177.51 (13)	N2''—C3''—C1'''—C2'''	9.4 (2)

The temperature of the crystal was controlled using an Oxford Cryosystems Cryostream Cooler (Cosier & Glazer, 1986). Data were collected over a hemisphere of reciprocal space, by a combination of three sets of exposures. Each set had

a different φ angle for the crystal and each exposure of 10 s covered 0.3° in ω . The crystal-to-detector distance was 5.01 cm. Coverage of the unique set was over 91% complete to at least 28° in θ . Crystal decay was monitored by repeating the initial frames at the end of the data collection and analysing the duplicate reflections, and was found to be negligible. H atoms were added at calculated positions and refined using a riding model. Anisotropic displacement parameters were used for all non-H atoms; for H atoms, $U(\text{H}) = 1.2U_{\text{eq}}(\text{C or O})$ or $1.5U_{\text{eq}}(\text{C})$ for methyl groups.

Data collection: *SMART* (Siemens, 1994a). Cell refinement: *SAINT* (Siemens, 1995). Data reduction: *SAINT*. Program(s) used to solve structure: *SHELXTL/PC* (Siemens, 1994b). Program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997). Molecular graphics: *SHELXTL/PC*. Software used to prepare material for publication: *SHELXTL/PC*.

We wish to acknowledge the use of the EPSRC's Chemical Database Service at Daresbury Laboratory (Fletcher *et al.*, 1996) for access to the Cambridge Structural Database (Allen & Kennard, 1993).

Supplementary data for this paper are available from the IUCr electronic archives (Reference: CF1247). Services for accessing these data are described at the back of the journal.

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Acta Cryst. (1998). **C54**, 1125–1127

9,10-Dihydro-1,4-dimethoxy-5,8-dimethyl-*cis*-9,10-diphenyl-9,10-anthracenediol, a Crowded Planar 9,10-Dihydroanthracene

CLAUS KRIEGER, ALEXANDER R. WARTINI AND FRANZ A. NEUGEBAUER

Arbeitsgruppe Organische Chemie, Max-Planck-Institut für Medizinische Forschung, Jahnstraße 29, D-69120 Heidelberg, Germany. E-mail: krieger@mixi.mpimf-heidelberg.mpg.de

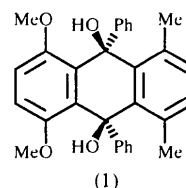
(Received 8 December 1997; accepted 23 February 1998)

Abstract

The crystal structure determination of the title compound, $\text{C}_{30}\text{H}_{28}\text{O}_4$, a key intermediate in the preparation of highly functionalized anthracenes, establishes the geometry at the central 9,10-positions to be *cis*. Owing to the buttressing effect of the methoxy and methyl groups in the *peri*-1,4,5,8-positions on the central 9,10-substituents, the 9,10-dihydroanthracene framework adopts a planar arrangement and the *cis*-9,10-phenyl groups assume a perpendicular conformation with regard to this plane.

Comment

The title compound, (1), and the corresponding *trans* isomer, which exhibit different reactivities, are key intermediates in the preparation of specifically functionalized anthracenes (Wartini, 1997). Therefore, an unambiguous elucidation of their geometry was required.



The crystal structure determination establishes a *cis* geometry for (1), and reveals that the tricyclic skeleton is essentially planar (Fig. 1). This is substantiated by the interplanar angle of $172.7(5)^\circ$ between the C1–C4, C4a, C9a and C5–C8, C8a, C10a arene planes, and by the relevant C10–C4a–C9a–C9 and C9a–C4a–C10–C10a torsion angles of $0.7(2)$ and $-1.9(2)^\circ$, respectively. In the constrained almost-planar 'twisted pseudo-boat'-shaped 1,4-cyclohexadiene ring, the bridging C9 and C10 atoms deviate from the C4a, C10a, C8a, C9a plane by only $0.025(1)$ and $0.016(1)$ Å, respectively. The angles at the central ring C9 and C10 atoms are distorted from a tetrahedral arrangement, *e.g.* at C9, C8a–